

# Topic 1: Crystal Precipitation

1

in batch crystallizer

Ostwald Ripening process

Batch crystallization - used in making many products...  
- big business in chemical industry  
pharmaceutical

Very often, crystals of certain sizes are desired.

This is done in a crystallizer, based on Ostwald Ripening:

- mix crystals of any sizes (produced by precipitation from solution) into appropriate solvent
- keep mixed, usually by stirring, to promote Ostwald Ripening:  
Larger crystals grow at the expense of smaller ones (via diffusion)  
"thermodynamic capitalism"!  
slower and slower...
- when crystals are about the desired size, remove them (often by drying out solvent.)

There are various models of the process (involving PDEs)

We will discuss a simple but successful one, involving ODEs.

# A <sup>simple</sup> model for Ostwald Ripening

- Assume crystals of same shape, say cubes, of size  $x$ , for simplicity

- A solute-solvent equilibrium is governed by a



solubility curve

$c$  = solute concentration

$c_{crit}$  = saturation concentration

= max the system can hold in solution at  $T, P$

if  $c > c_{crit}$  then the excess precipitates out as solid (crystals)

if  $c < c_{crit}$  then crystals dissolve into the solvent

- Gibbs-Thomson relation:  $c_{crit} = c^* e^{\frac{\Gamma}{x}}$

$$\Gamma = \frac{4\sigma v}{RT}$$

$\sigma$  = surface energy  $[\frac{J}{m^2}]$

$v$  = molar volume of solute  $[\frac{m^3}{mol}]$

$R$  = gas constant  $[\frac{J}{mol \cdot K}]$

$T$  = temperature  $[K]$

$c^*$  = saturation conc, at  $\infty$  dilution

so, the larger the size  $x$ , the lower the triggering value

so, larger crystals grow easier! "thermodynamic capitalism!"

- solute concentration in solution:  $c = c_0 + \mu(x^*)^3 - \mu x^3$

$x^*$  = initial size (at time zero)

$c_0$  = initial concentration of solute in solution

- $c^*, \Gamma, \mu, c_0, x^*$ : given constants (parameters)

$x(t)$  : time-dependant unknown

need a law (rule) for how  $x(t)$  evolves in time...

Empirical (phenomenological) kinetic Law for  $x(t)$ :

$$\frac{dx}{dt} = \begin{cases} k_g (c - c_{\text{crit}})^g & \text{if } c > c_{\text{crit}} \\ -k_d (c_{\text{crit}} - c)^d & \text{if } c < c_{\text{crit}} \end{cases}$$

for some parameters  $k_g, k_d > 0$ ,  $1 \leq g, d \leq 2$ .

So, if  $c > c_{\text{crit}}$  then  $\frac{dx}{dt} > 0$  so  $x$  grows  
 if  $c < c_{\text{crit}}$   $\frac{dx}{dt} < 0$  decays  
 if  $c = c_{\text{crit}}$   $\frac{dx}{dt} = 0$  no change

This is an ODE for  $x(t)$ , of 1<sup>st</sup> order

$$\frac{dx}{dt} = G(x)$$

with initial condition:  $x(0) = x^* = \text{initial size}$ .

For simplicity (and convenience) we take  $g = d = 1$

and  $k_g = k_d =: k$

in which case  $G(x)$  simplifies to

$$\begin{aligned} G(x) &= k(c - c_{\text{crit}}) \\ &= k \left[ c_0 + \mu (x^*)^3 - \mu x^3 - c^* e^{\frac{c}{x}} \right] \end{aligned}$$

# Evolution model for crystals of $N$ sizes

4

Assume initially there are  $N$  different sizes

$$0 < x_1^* < \dots < x_N^*$$

They evolve according to our kinetic law to sizes  $x_1(t), \dots, x_N(t)$  governed by

$$(IVP) \begin{cases} \frac{dx_j}{dt} = G_j(x_1, \dots, x_N) = k [c(t) - c_{crit,j}] \\ x_j(0) = x_j^* \end{cases} \quad j=1, 2, \dots, N$$

$$\text{where } c(t) = c_0 + \sum_{j=1}^N \mu_j (x_j^*)^3 - \sum_{i=1}^N \mu_i (x_i(t))^3$$
$$= \underbrace{c_1}_{\text{(constant)}} - \sum_{i=1}^N \mu_i x_i(t)^3$$

$$c_{crit,j} = c^* e^{\frac{\Gamma}{x_j}} \quad , \quad j=1:N$$

This is an Initial Value Problem (IVP) for a system of  $N$  <sup>first order</sup> ODEs,

Very nonlinear,  $x_j$ 's appear in  $x_j^3$  and  $e^{\Gamma/x_j}$

In vector notation, setting  $\vec{X} = \begin{bmatrix} x_1 \\ \vdots \\ x_N \end{bmatrix}$ ,  $\vec{G} = \begin{bmatrix} G_1 \\ \vdots \\ G_N \end{bmatrix}$ ,  $\vec{X}^* = \begin{bmatrix} x_1^* \\ \vdots \\ x_N^* \end{bmatrix}$

$$(IVP) \quad \frac{d\vec{X}}{dt} = \vec{G}(\vec{X}), \quad \vec{X}(0) = \vec{X}^*$$

Simplest case: Single size crystals:  $N=1$

Consider crystals of a single size  $x(t)$  evolving from initial size  $x^*$ ; cubic shape

$$(IVP) \quad \frac{dx}{dt} = G(x), \quad x(0) = x^*$$

$$\text{where } G(x) = k [c - c_{\text{crit}}] = k [c_1 - \mu x^3 - c^* e^{\frac{\Gamma}{x}}]$$

$$\text{with } c(t) = \underbrace{c_0 + \mu(x^*)^3}_{c_1} - \mu x(t)^3, \quad c_{\text{crit}} = c^* e^{\frac{\Gamma}{x(t)}}$$

$$= c_1 - \mu x^3$$

Note: time does not appear explicitly, such ODEs are called autonomous.

Equilibria are constant (steady-state) solutions of an ODE  $\frac{dx}{dt} = G(x)$

so must satisfy  $G(x) = 0$ , so they are zeros (roots) of  $G(x)$ .

Are there any? how many?